

## Calculating Specific Heat Worksheet Answers

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### Calculating Specific Heat Worksheet Answers

Specific Heat Worksheet Name (in ink):  $C = q/m\Delta T$ , where  $q$  = heat energy,  $m$  = mass, and  $T$  = temperature Remember,  $\Delta T = (T_{\text{final}} - T_{\text{initial}})$ . Show all work and proper units. Answers are provided at the end of the worksheet without units. 1. A 15.75-g piece of iron sorbs 1086.75 joules of heat energy, and its temperature changes from 25 0 1750C.

### Specific Heat Wksht20130116145212867

Answers to Worksheet # 17 Calculating Heat The specific heat capacity ( $c$ ) of a substance is the amount of heat required to raise the temperature of 1 gram of a substance by 1 K. Units are in  $J/g\cdot K$  or  $J/g\cdot ^\circ C$ . The molar heat capacity ( $C$ ) of a substance is the amount of heat required to raise the temperature of

### Calculating Heat - University of Florida

Notes: Rules of series and parallel circuits are very important for students to comprehend. However, a trend I have noticed in many students is the habit of memorizing rather than understanding these rules. Students will work hard to memorize the rules without really comprehending why the rules are true, and therefore often fail to recall or apply the rules properly.

### Series-Parallel DC Circuits Worksheet - DC Electric Circuits

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The particle model describes matter in a very specific way. It describes four important aspects of matter: All matter is made up of particles that are incredibly small - much too small to see with the naked eye. The particles can be atoms or combinations of atoms that are bonded. There are forces between the particles.

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