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course. (a) 71 mL to 73 mL (b) 8.22 m to
8.24 m (c) 4.54×10^{-5} g to 4.56×10^{-5}
g 1.3. Mastering ...

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metals. nonmetals. metalloid. when the
elements are arranged in order of
increasing atomic n.... one of a class of
elements that are good conductors of

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heat an.... an element that tends to be a poor conductor of heat and elect....

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CHAPTER 6 THERMOCHEMISTRY. 6.15

Recall that the work in gas expansion is equal to the product of the external, opposing pressure and the change in volume. (a) $w = -P\Delta V$. $w = -(0)(5.4 - 1.6)L = 0$. (b) $w = -P\Delta V$. $w = -(0.80 \text{ atm})(5.4 - 1.6)L = -3.0 \text{ L}\cdot\text{atm}$. To

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eight 3. nine 4. accepted 5. Dmitri
Mendeleev 6. atomic mass 7. elements
8. atomic number 9. Henry Moseley 10.
protons 11. periodic law

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Chemistry Answer Key

Ch 6 Study Guide answers

(a) (b) Answer is one of the following.
A/yellow: mass = 65.14 kg, volume = 3.38 L, density = 19.3 kg/L, likely identity = gold. B/blue: mass = 0.64 kg, volume = 1.00 L, density = 0.64 kg/L, likely identity = apple.

Answer Key Chapter 1 - Chemistry

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things into groups based on their chemical composition Chapter 6 chemistry of living things answer key. Carbohydrates (starches and sugars) are composed of carbon, hydrogen, and oxygen. Carbohydrates are the main energy source for living things Chapter 6 chemistry of living things answer key.

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