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Add varying amounts of a chemical to a beaker of water to create a solution, observe that the chemical dissolves in the water at first, and then measure the concentration of the solution at the saturation point. Either

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potassium nitrate or sodium chloride can be added to the water, and the temperature of the water can be adjusted.

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Solubility and Temperature Answer

Key Vocabulary:

concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute,

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solution, solvent Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking.]

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Solubility refers to the _____ of solute that can dissolve in a certain volume or mass of solvent, at a certain temperature. Solubility & Temperature Gizmo Quiz DRAFT 9th - 12th grade

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Gizmo Warm-up A solution generally consists of two parts, a solute that is dissolved and a solvent that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the Solubility and Temperature Gizmo™, you will study how temperature affects

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how much solute will
dissolve in a solution.

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In the Solubility and
Temperature Gizmo™,
you will study how
temperature affects
how much solute will
dissolve in a solution.
To begin, check that
Potassium nitrate is
selected and the Temp.
of the water is 20 °C.

Click OK 1. In this

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solution, what is the solute?

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Key Vocabulary:

concentration,
dissolve, homogeneous
mixture, solubility,
solubility curve, solute,
solution, solvent

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students thinking.

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concentration of the

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