

## Solutions Notes

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**Solutions Notes**  
1. A solution is a homogeneous mixture of two or 9. more chemically non-reacting substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2. A solution may be classified as solid, liquid or a gaseous solution. 3.

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Chemistry Notes for class 12 Chapter 2 Solutions. Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution.

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Dilutions <u></u>Chemists will often store solutions in a more concentrated form to save on storage space and the ease of making new solutions. These types of solutions are called stock solutions. Virtually an infinite number of solutions of varying concentrations can be made from a single stock solution, simply by adding more solvent.

**Solutions Notes**  
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1. Acidic buffer solution: These are solutions that have a pH below 7 and contain a weak acid and one of its salts. For example, a mixture of acetic acid and sodium acetate acts as a buffer solution with a pH of about 4.75. 2. Alkaline (Basic) buffer solution: These are solutions that have a pH above 7 and contain a weak base and one of its ...

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